

Locomotor inhibitory Activity of Some Murrayanine-Chalcone based 2, 3-dihydrobenzo[b] [1, 4] thiazepine Derivatives: Exploring Anxiolytic Potentials

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Abstract

The seven-membered ring candidate benzothiazepine is a well-known scaffold that finds the majority of applications in hypnotic, sedative, anxiolytic, and other CNS associated therapeutics. Marketed drugs like diltiazem, clemiazem, and several other emerging molecules are the best example of this scaffold prototype. Murrayanine is the highly explored carbazole compound present in Indian curry plant *Murraya koenigii* L. (Rutaceae). A lot of heterocyclic hybrids have been synthesized and their pharmacotherapeutic significance has been revealed by our research group.

The current study involved the development of a number of seven-membered containing heterocycle; murrayanine-2-(substituted)-phenyl-2,3-dihydrobenzo[b] [1,4] thiazepine derivatives from murrayanine-chalcone, a previously reported component by our research group by cyclization reaction. The locomotor inhibitory activity was explored in Swiss albino rat which may be translated with anti-anxiety or hypnotic effects by the fabricated molecules. The compound 3g displayed the highest inhibition (60.17%) of the locomotor activity followed by compounds 3h and 3f. The analytical tools helped to determine the structure of proposed compounds. The position, number, and the type of substituent exerted a decisive role in mediating the biological activity. However, a crystal-clear SAR cannot be predicted from this study. The mode of action of the benzothiazepine analogs may be considered by enhancing the effect of GABA neurotransmitter at GABAA receptor to produce anxiolysis and hypnosis. The present research will definitely attract researchers across the world by providing clues for rational designing of natural product-based inhibitors with multifarious pharmacological activities.

Keywords: Murrayanine; Chalcone; Benzothiazepine; Heterocycle; Locomotor; *Murraya koenigii*

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Introduction

The six-membered ring fused seven-membered ring candidates such as benzodiazepine, benzothiazepine, and benzoxazepine are the well-known scaffold that finds the majority of applications in hypnotic, sedative, anxiolytic, and other CNS associated therapeutics. [1] Benzothiazepines are the class of drugs which have multifarious pharmacological potentials such as anti-inflammatory [2], anti-retroviral [3], anti-cancer [4], anti-bacterial [5], anti-fungal [6], anti-malarial [7], anti-viral [8], anti-convulsant [9], anti-diabetic [10], anti-depressant [11], anti-oxidant [12], anti-hypertensive [13], hypolipidemic [14], etc. Marketed drugs like diltiazem, clemizem, and several other emerging molecules are the best example of this scaffold prototype.

Murraya koenigii L. is also known as Indian curry plant, belonging to the family Rutaceae. [15] Ethnopharmacological importance of the plant extracts of the root, leaf, and stem bark include febrifuge, carminative, anti-helminthic, stomachic, purgative, astringent, etc. [16] Scientific studies over the years have reported the presence of therapeutically active carbazole moieties such as mahanine, bismahanine, isomahanine, euchrestine B, bispyrayafoline, bismurrayafoline, koenimbine, O-methylmahanine, O-methylmurrayamine A, mahaninebicine, mahaninebine, and murrayanine which displayed potent anti-oxidant, anti-bacterial, anti-fungal, anti-ulcerogenic, and immunomodulatory activities. [17] Murrayanine is the highly explored compound of this series and a lot of heterocyclic hybrids have been synthesized and their pharmacotherapeutic significance has been revealed by our research group. [18-25]

The current study involved the development of a number of seven-membered containing heterocycle; murrayanine-2-(substituted)-phenyl-2,3-dihydrobenzo[b] [1,4] thiazepine derivatives from murrayanine-chalcone, a previously reported component by our research group by cyclization reaction. The locomotor inhibitory activity was explored in Swiss albino rat which may be translated with anti-anxiety or hypnotic effects by the fabricated molecules.

Materials and Methods

Chemical and Instrumentation

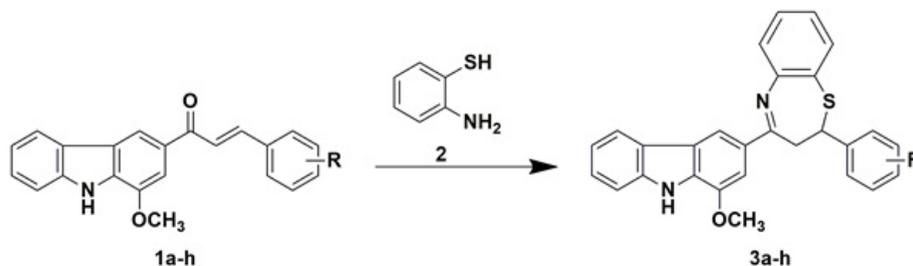
The chemical reaction was initiated from "Murrayanine-Chalcone", a chemical component previously reported by our research group. The chemical reagents and solvents of analytical grade were procured from Merck, Sigma-Aldrich, and HiMedia. The progress of the reaction was monitored by Merck Pre-coated silica gel G TLC plates. The CHN analyses were performed using PerkinElmer 2400 model Elemental Analyzer. The FT-IR spectra were recorded on Shimadzu® IRAffinity-1 infrared spectrometer using KBr methods and expressed in cm^{-1} . The mass spectra were recorded on MICROMASS Q-TOF instrument. The proton (^1H) NMR was recorded using Bruker Avance-II instrument using tetramethylsilane (TMS) as the internal standard and expressed in ppm relative to the internal standard.

Animals

The experiment was performed after receiving approval from the Department Ethical Committee (DEC) and CPCSEA (1389/a/10/CPCSEA) on Swiss albino rat (6 rats per group) of age 5-6 weeks and weight 180-280 g range. The animals were housed under good hygienic and controlled conditions where 24–25°C temperature, humidity 50–60%, and 12 hr light and dark cycle was maintained. The rats were provided free access to water and standard rodent pellets.

Synthesis of target compounds

The fabrication of 2, 3-dihydrobenzo [b] [1, 4] thiazepine derivatives (3a-h) involved the transformation of murrayanine-chalcone (1a-h), the compound reported previously by our research group, by reacting with 2-aminothiophenol (2). The chemical reaction involved immediate modification of the carbonyl segment into the closed ring benzothiazepine form. The outline of chemical transformation is described in Scheme 1.



Scheme 1: The synthesis of benzothiazepine derivatives from the murrayanine-chalcones.

Synthetic protocol for 4-(1-methoxy-9H-carbazol-3-yl)-2-(substituted)-phenyl-2, 3-dihydrobenzo[b] [1,4] thiazepine (3a-h). The murrayanine-chalcone derivatives (1a-h) (0.1 M) was dissolved in ethanol media and reacted with equimolar quantity of 2-aminothiophenol (2) in presence of 7-8 drops of glacial acetic acid. The content was refluxed for 8-10 hrs and the progress of the reaction was inspected using TLC. After validating the completion of reaction, crushed ice was added into the reaction mixture to acquire the crude solid product. The product was filtered using Buchner's funnel, washed carefully with cold water, and recrystallized from raw ethanol to achieve pure compound.

2-(2-fluorophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2, 3-dihydrobenzo[b] [1, 4] thiazepine (3a) 37% yield; FTIR (KBr) ν (cm⁻¹): 3247 (-NH), 3152 (C-H, aromatic), 1684 (C = N, aromatic), 1621 (C = C, aromatic), 1592 (-NH, bending), 1326 (C-N), 1258 (C-O), 1197 (C-F); ¹H NMR (δ , ppm, CDCl₃): 10.14 (9, 1H), 7.0-8.4 (Aromatic, 14H), 3.91 (13, 1H), 3.89 (1, 3H), 2.18 (12, 1H); MS: M⁺ 452. Anal. Calcd. For C₂₈H₂₁FN₂OS: C, 74.31; H, 4.68; N, 6.19. Found: C, 73.92; H, 4.23; N, 5.88.

2-(4-fluorophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3b) 52% yield; FTIR (KBr) ν (cm⁻¹): 3295 (-NH), 3129 (C-H, aromatic), 1644 (C = N, aromatic), 1638 (C = C, aromatic), 1554 (-NH, bending), 1314 (C-N), 1225 (C-O), 1150 (C-F); ¹H NMR (δ , ppm, CDCl₃): 10.19 (9, 1H), 7.1-8.6 (Aromatic, 14H), 3.88 (13, 1H), 3.81 (1, 3H), 2.13 (12, 1H); MS: M⁺ 452. Anal. Calcd. For C₂₈H₂₁FN₂OS: C, 74.31; H, 4.68; N, 6.19. Found: C, 73.81; H, 4.15; N, 5.82.

2-(2-iodophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3c) 45% yield; FTIR (KBr) ν (cm⁻¹): 3274 (-NH), 3135 (C-H, aromatic), 1661 (C = N, aromatic), 1617 (C = C, aromatic), 1579 (-NH, bending), 1331 (C-N), 1279 (C-O), 691 (C-I); ¹H NMR (δ , ppm, CDCl₃): 10.12 (9, 1H), 7.0-8.6 (Aromatic, 14H), 3.93 (13, 1H), 3.84 (1, 3H), 2.17 (12, 1H); MS: M⁺ 560. Anal. Calcd. For C₂₈H₂₁IN₂OS: C, 60.01; H, 3.78; N, 5.00. Found: C, 59.26; H, 3.18; N, 4.43.

2-(4-iodophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3d) 61% yield; FTIR (KBr) ν (cm⁻¹): 3232 (-NH), 3148 (C-H, aromatic), 1673 (C = N, aromatic), 1624 (C = C, aromatic), 1565 (-NH, bending), 1309 (C-N), 1262 (C-O), 658 (C-I); ¹H NMR (δ , ppm, CDCl₃): 10.16 (9, 1H), 7.0-8.4 (Aromatic, 14H), 3.96 (13, 1H), 3.79 (1, 3H), 2.11 (12, 1H); MS: M⁺ 560. Anal. Calcd. For C₂₈H₂₁IN₂OS: C, 60.01; H, 3.78; N, 5.00. Found: C, 59.34; H, 3.27; N, 4.39.

2-(4-bromophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3e) 49% yield; FTIR (KBr) ν (cm⁻¹): 3261 (-NH), 3127 (C-H, aromatic), 1651 (C = N, aromatic), 1631 (C = C, aromatic), 1583 (-NH, bending), 1335 (C-N), 1243 (C-O), 637 (C-Br); ¹H NMR (δ , ppm, CDCl₃): 10.11 (9, 1H), 7.0-8.4 (Aromatic, 14H), 3.87 (13, 1H), 3.73 (1, 3H), 2.21 (12, 1H); MS: M⁺ 512, M+2 514. Anal. Calcd. For C₂₈H₂₁BrN₂OS: C, 65.50; H, 4.12; N, 5.46. Found: C, 64.76; H, 3.87; N, 5.08.

4-(1-methoxy-9H-carbazol-3-yl)-2-(2-(trifluoromethyl) phenyl)-2, 3-dihydrobenzo[b] [1, 4] thiazepine (3f) 33% yield; FTIR (KBr) ν (cm⁻¹): 3258 (-NH), 3166 (C-H, aromatic), 1649 (C = N, aromatic), 1611 (C = C, aromatic), 1598 (-NH, bending), 1317 (C-N), 1251 (C-O), 1118 (C-F); ¹H NMR (δ , ppm, CDCl₃): 10.21 (9, 1H), 7.2-8.4 (Aromatic, 14H), 3.83 (13, 1H), 3.75 (1, 3H), 2.14 (12, 1H); MS: M⁺ 502. Anal. Calcd. For C₂₉H₂₁F₃N₂OS: C, 69.31; H, 4.21; N, 5.57. Found: C, 68.57; H, 3.83; N, 5.18.

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2-(3,5-bis(trifluoromethyl)phenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3g) 40% yield; FTIR (KBr) ν (cm^{-1}): 3244 (-NH), 3139 (C-H, aromatic), 1672 (C = N, aromatic), 1640 (C = C, aromatic), 1559 (-NH, bending), 1322 (C-N), 1216 (C-O), 1135 (C-F); ^1H NMR (δ , ppm, CDCl_3): 10.15 (9, 1H), 7.2-8.5 (Aromatic, 13H), 3.86 (13, 1H), 3.80 (1, 3H), 2.19 (12, 1H); MS: M^+ 520. Anal. Calcd. For $\text{C}_{29}\text{H}_{20}\text{F}_4\text{N}_2\text{OS}$: C, 66.91; H, 3.87; N, 5.38. Found: C, 65.82; H, 3.59; N, 4.93.

2-(2,4-dichloro-5-fluorophenyl)-4-(1-methoxy-9H-carbazol-3-yl)-2,3-dihydrobenzo[b] [1,4] thiazepine (3h) 72% yield; FTIR (KBr) ν (cm^{-1}): 3277 (-NH), 3163 (C-H, aromatic), 1656 (C = N, aromatic), 1607 (C = C, aromatic), 1564 (-NH, bending), 1312 (C-N), 1234 (C-O), 1141 (C-F), 753 (C-Cl); ^1H NMR (δ , ppm, CDCl_3): 10.13 (9, 1H), 7.1-8.7 (Aromatic, 12H), 3.97 (13, 1H), 3.89 (1, 3H), 2.17 (12, 1H); MS: M^+ 520, $M+2$ 522. Anal. Calcd. For $\text{C}_{28}\text{H}_{19}\text{FN}_2\text{OS}$: C, 64.50; H, 3.67; N, 3.64. Found: C, 64.11; H, 3.24; N, 3.17.

Acute toxicity studies

Acute toxicity study was performed to determine the *in vivo* safety profile of the experimental compounds and to approximate the dose which will express the highest promising biological activity with no manifestation of any toxicity signs and symptoms. The compounds were injected at gradually escalating quantity ranging from 10 mg/kg to 100 mg/kg. The dose was calculated based on the death of 50% animals.

Inhibition of locomotor activity: Exploring anti-anxiety effect

Using an actophotometer, the locomotor activity was studied comprehensively with little modification. For the experiment, each animal was placed separately in the actophotometer system containing photocells. For each experimental animal, the basal activity score was calculated after 10 and 20 min of drug administration. The locomotor activity on each rat was retested for 10 min. The difference in the locomotor activity values was recorded before and after the drug treatment. Finally, the reduction in locomotor activity was calculated and expressed in percentage.

Statistical treatment

The obtained data were compared with the control group and analyzed statistically by one-way ANOVA method followed by Dunnett's multiple comparisons test. A value $P < 0.01$ was considered statistically significant.

Result and Discussion

Chemistry

The alteration of chalcone into the seven-membered heterocycle benzothiazepine was supported by sophisticated analytical techniques. The conversion of the ketonic C = O moiety of the murrayanine-chalcone into the cyclic form was strongly confirmed by the disappearance of the characteristic ketonic peak in the FT-IR spectra, which appeared previously at $1670\text{-}1780\text{ cm}^{-1}$. The amide components present in carbazole and benzothiazepine moiety were largely characterized by absorption frequencies at (stretching) $3232\text{-}3298\text{ cm}^{-1}$ and $1554\text{-}1598$ (bending) cm^{-1} , respectively.

The features of the aromatic ring were noticed by stretching of C-H and C = C components at $3127\text{-}3166\text{ cm}^{-1}$ and $1607\text{-}1640\text{ cm}^{-1}$, respectively. The ^1H -NMR studies demonstrated the discrete attributes of the prepared benzothiazepine molecules. The proton associated with the carbazole nitrogen appeared principally at 10 ppm. Chiefly, the aromatic ring hydrogens were positioned in the spectral range of 7.0-8.6 ppm. The protons of the methoxy group were perceived at 3.8-3.9 ppm in the NMR spectra. The mass spectra of the compounds revealed that base peaks matched exactly with the theoretical molecular mass of the molecules. The isotope forms of the chlorine and bromine were identified by the base peak + 2 molecular mass. Additionally, a few fragment peaks were noticed in the range of m/z 100-200. The determined ratios of carbon, hydrogen, and nitrogen illustrated the probable composition of the derivatives which indeed supported the fabrication of the molecules. These above spectroscopical results authenticate the formation of seven-membered heterocycles from murrayanine-chalcone.

Determination of LD₅₀ value

The benzothiazepine derivatives were found to be relatively safe as no symptoms and signs of toxicity were detected over the administered dose range of 10-100 mg/kg b.w. For carrying out the experiment, a fixed dose of 30 mg/kg b.w. was used for determining the locomotor inhibitory activity in Swiss albino rats.

Locomotor inhibitory activity

The experimental compounds demonstrated moderate to fairly high inhibitory effect in the treated animals by acting directly on the CNS interface. All the molecules displayed a nearly same degree of inhibitory potentials. The compound 3g displayed the highest inhibition (60.17%) of the locomotor activity, followed by compounds 3h and 3f which displayed inhibition of 57.97% and 54.98%, respectively. On focusing the structure-activity-relationships (SARs), it was observed that the number, position, and the type of substituent has a critical role in imparting pharmacological activity. In fluorine substituents (3a and 3b), the para-position was found to be more prevalent for locomotion inhibitory effect. In contrast, the iodine substituents (3c and 3d), the *ortho*-position was found to be more privileged. The phenomenon observed in the case of benzothiazepine was entirely opposite with that of data obtained in the case of benzodiazepine. In the case of bromo-substituent (3e), a moderate inhibitory activity in the rats was detected (48.22%). However, none of the compounds exhibited better or higher pharmacological activity than the standard drug, benzodiazepine (diazepam). The lipophilicity may be believed to be the crucial factor in the reduced biological activity of some of the analogs. The fall in the activity may be explained on the basis that the drug molecules distributed to all compartments of the body and are available in the least quantity in the CNS interface.[26] In other ways, it may be enlightened that the formation of micelles of the molecules or binding to the amino acid residues may produce obstruction in crossing the biological barrier.[27] The mode of action of the benzothiazepine analogs may be considered by enhancing the effect of GABA neurotransmitter at GABAA receptor to produce anxiolysis and hypnosis. The mechanism may be supposed to be pretty similar to that of benzodiazepine (diazepam).

Group	R	Photocell count in 10 min	% inhibition	Photocell count in 20 min	% inhibition
Control*	-	402.8 ± 2.14	-	408.2 ± 2.36	-
Standard#	-	112.2 ± 0.29	72.15	103.6 ± 0.41	74.63
3a&	2-F	214.8 ± 1.72**	46.68	202.2 ± 2.84*	50.47
3b	4-F	205.4 ± 2.38*	49.01	193.4 ± 1.55*	52.63
3c	2-I	234.6 ± 1.63*	41.76	219.2 ± 1.37**	46.31
3d	4-I	244.6 ± 2.24*	39.28	234.2 ± 2.61**	42.63
3e	4-Br	226.2 ± 1.94**	43.85	211.4 ± 1.43**	48.22
3f	2-CF ₃	196.4 ± 1.17**	51.25	183.8 ± 1.76**	54.98
3g	3,5-CF ₃	173.6 ± 1.46**	56.91	162.6 ± 2.99*	60.17
3h	2,4-Cl; 5-F	185.6 ± 2.66*	53.93	171.6 ± 1.21**	57.97

Table 1: Locomotor inhibitory potential of benzothiazepine derivatives in rats.

*0.9% saline; #Benzodiazepine – 3 mg/kg b.w.; &Dose of 20 mg/kg b.w.;

**P < 0.01, *P < 0.05; Values expressed as mean ± SEM, from 6 rats.

Conclusion

The current research revealed the method to fabricate semi-synthetic heterocycles from natural product murrayanine having high limits of safety and efficacy. The compound 3g displayed the highest inhibition (60.17%) of the locomotor activity followed by compounds 3h and 3f. The analytical tools helped to determine the structure of proposed compounds. The position, number, and the type of

substituent exerted a decisive role in mediating the biological activity. However, a crystal-clear SAR cannot be predicted from this study. The mode of action of the benzothiazepine analogs may be considered by enhancing the effect of GABA neurotransmitter at GABAA receptor to produce anxiolysis and hypnosis. The present research will definitely attract researchers across the world by providing clues for rational designing of natural product-based inhibitors with multifarious pharmacological activities.

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